

Molecular modeling activities

Computational Materials Science and Technology Lab
CMAST Laboratory : www.afs.enea.it/project/cmast

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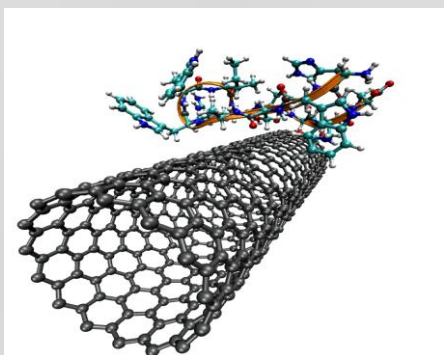
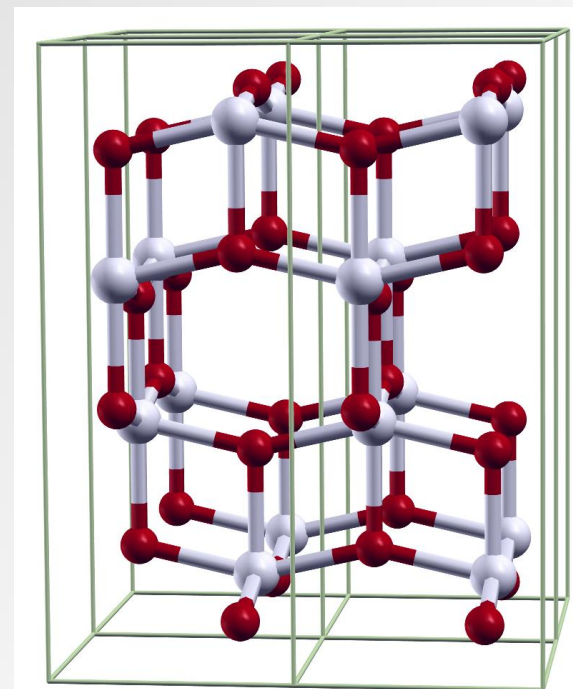
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Workshop: "Supercomputing, applicazioni e innovazioni: le attività scientifiche in ENEA supportate da CRESCO"
Roma, 11.7.2013

Molecular Dynamics simulations

MODEL

System Hamiltonian
Interaction between particles
Bonded and non bonded interactions
Forces on particles

$$H = \sum_{i=1}^N \frac{p_i^2}{2m_i} + \sum_{i<j}^N u(r_{ij}) + \dots + U_{\text{ext}}$$

$$\mathbf{F}_i = - \sum_{j=1, j \neq i}^N \frac{\partial u_{ij}(r)}{\partial \mathbf{r}_{ij}}$$

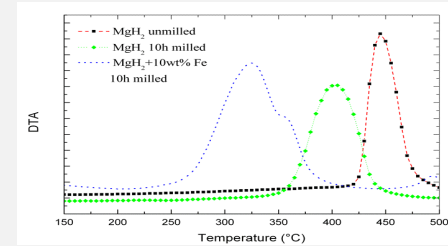
INTEGRATOR

Propagation through phase space
Finite difference schemes
Equation integrators
Numerical stability

$$\dot{p}_i = -\frac{\partial H}{\partial q_i}, \quad \dot{q}_i = \frac{\partial H}{\partial p_i}$$

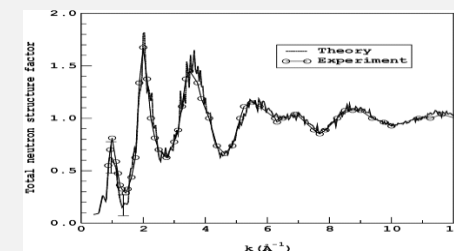
STATISTICAL ENSEMBLE

Thermodynamical conditions
Microcanonical ensemble
Canonical ensemble
Isothermal-isobaric ensemble

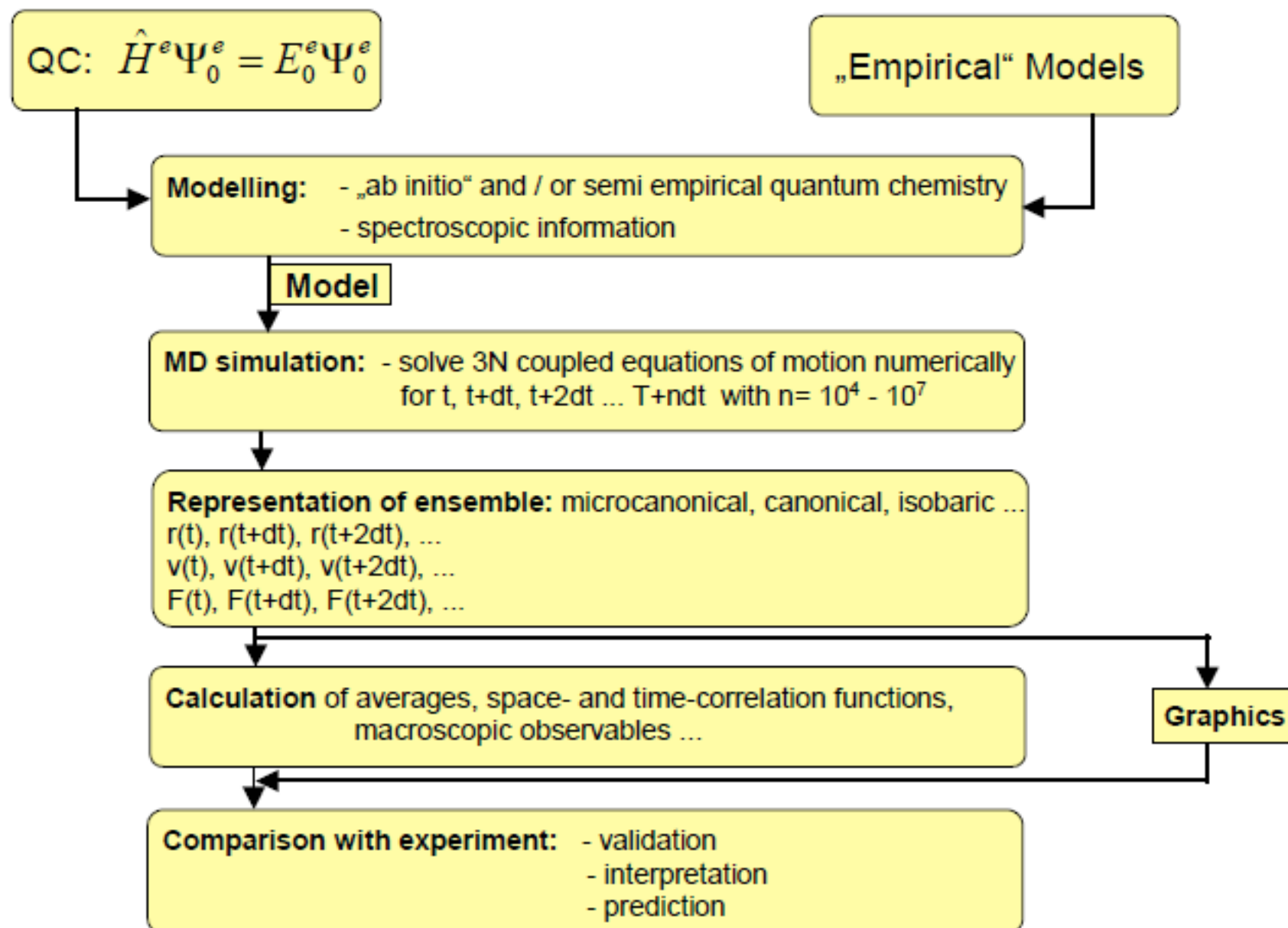


RESULTS

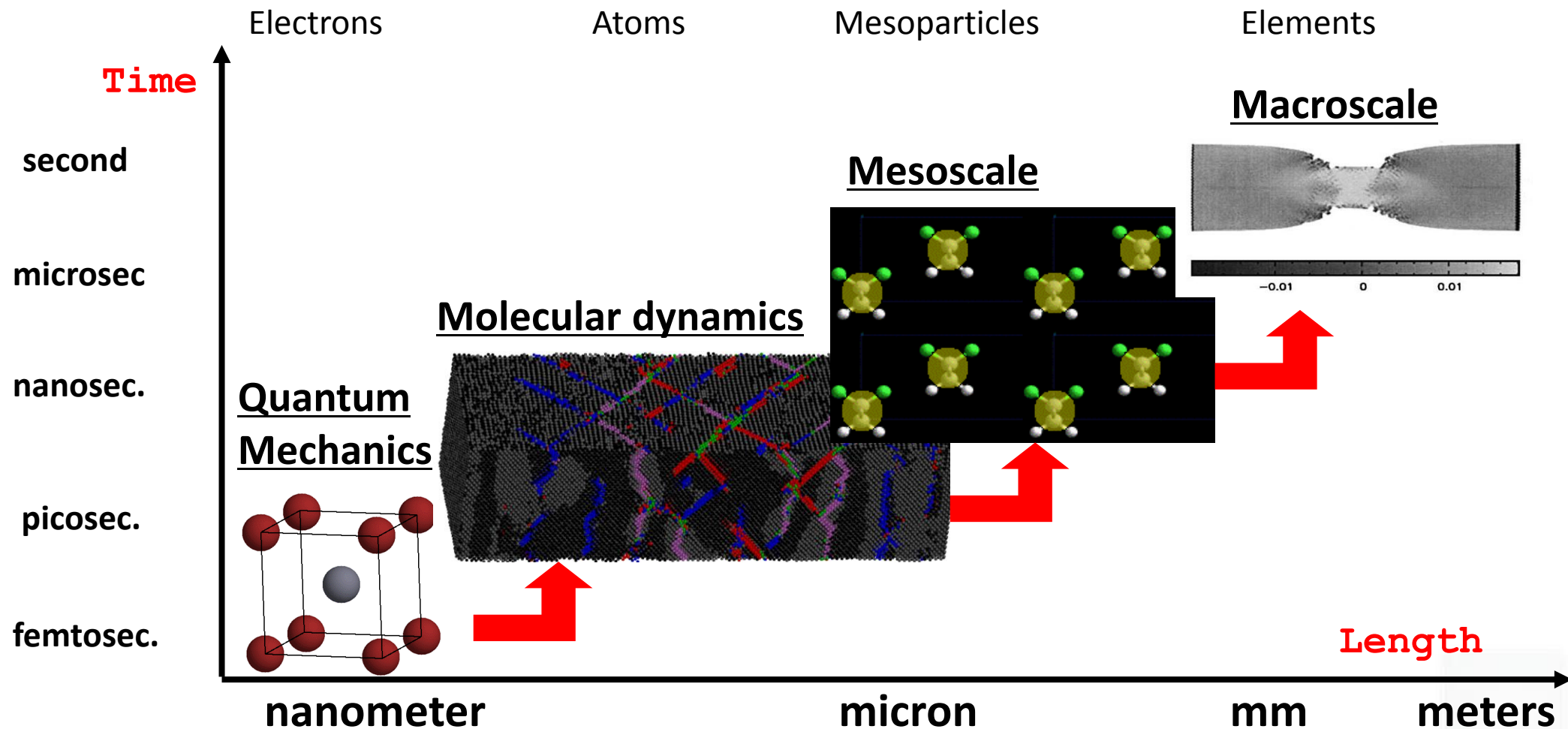
Thermodynamics and statistical mechanics
Internal energy, pressure, temperature
Response functions, correlation functions,
linear response theory



Flowchart of a Molecular Dynamics code



Multiscale modeling of materials

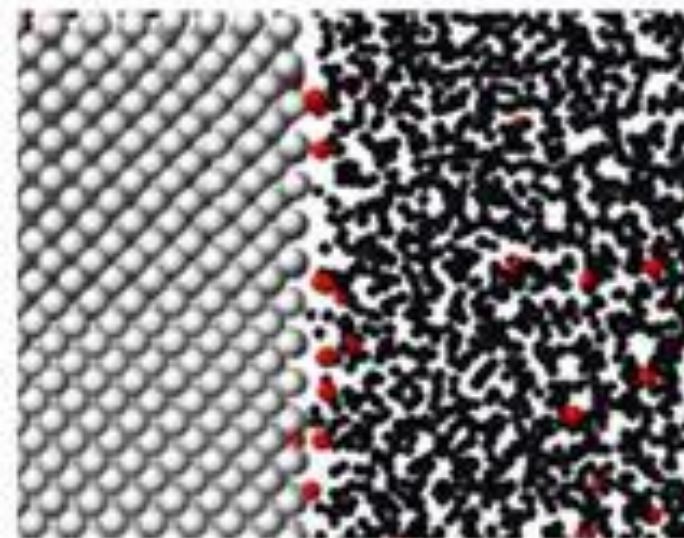


GOALS

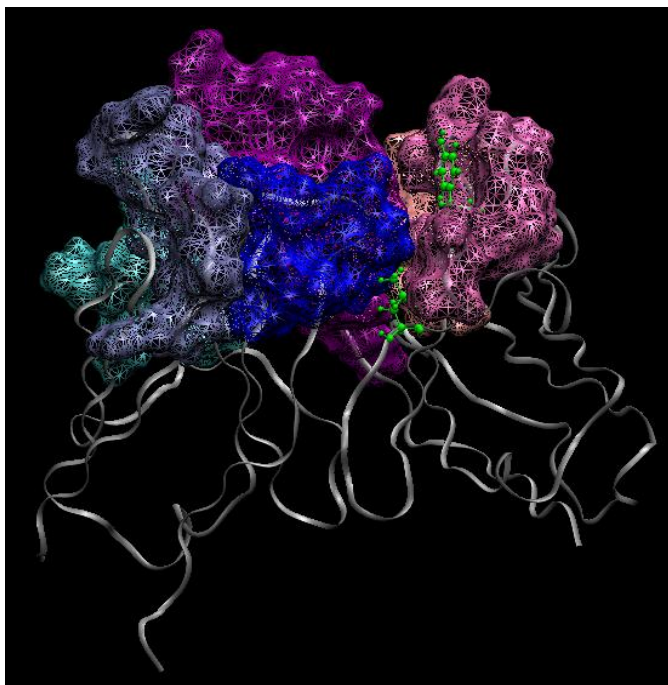
- To understand the molecular level origin of materials behaviour
- To predict the behaviour of materials
- To design new materials or devices with improved performance

Molecular modeling activities
in the field of

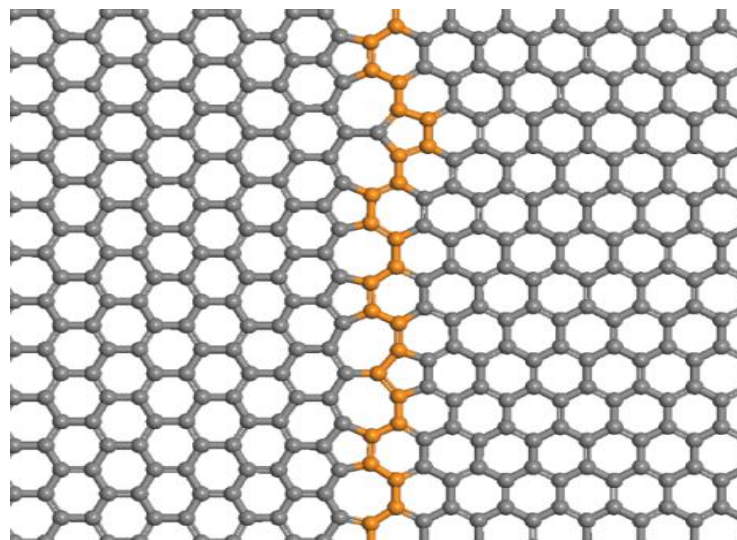
Materials



Biomolecules



Nanotechnologies

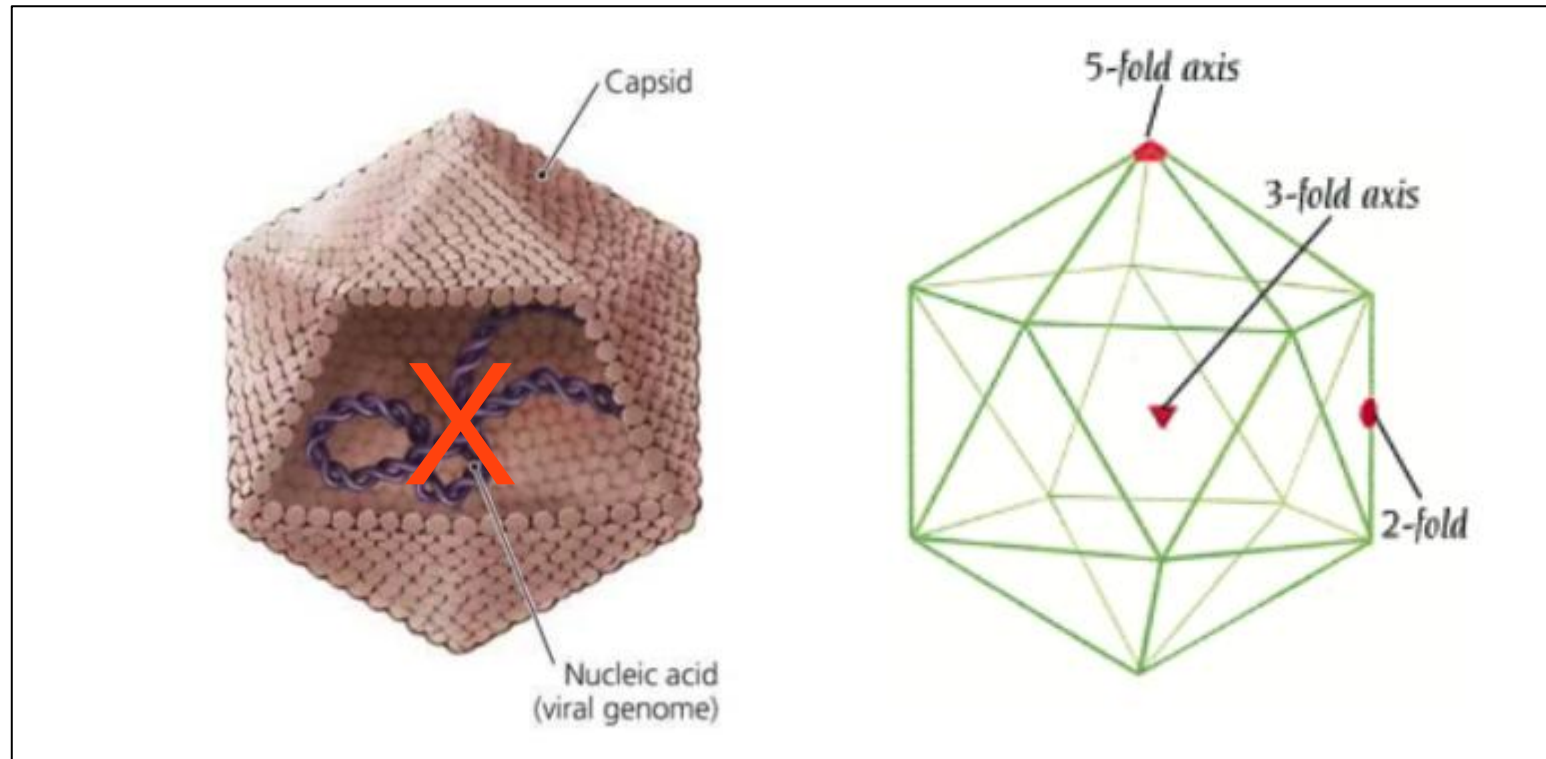


zig-zag edge

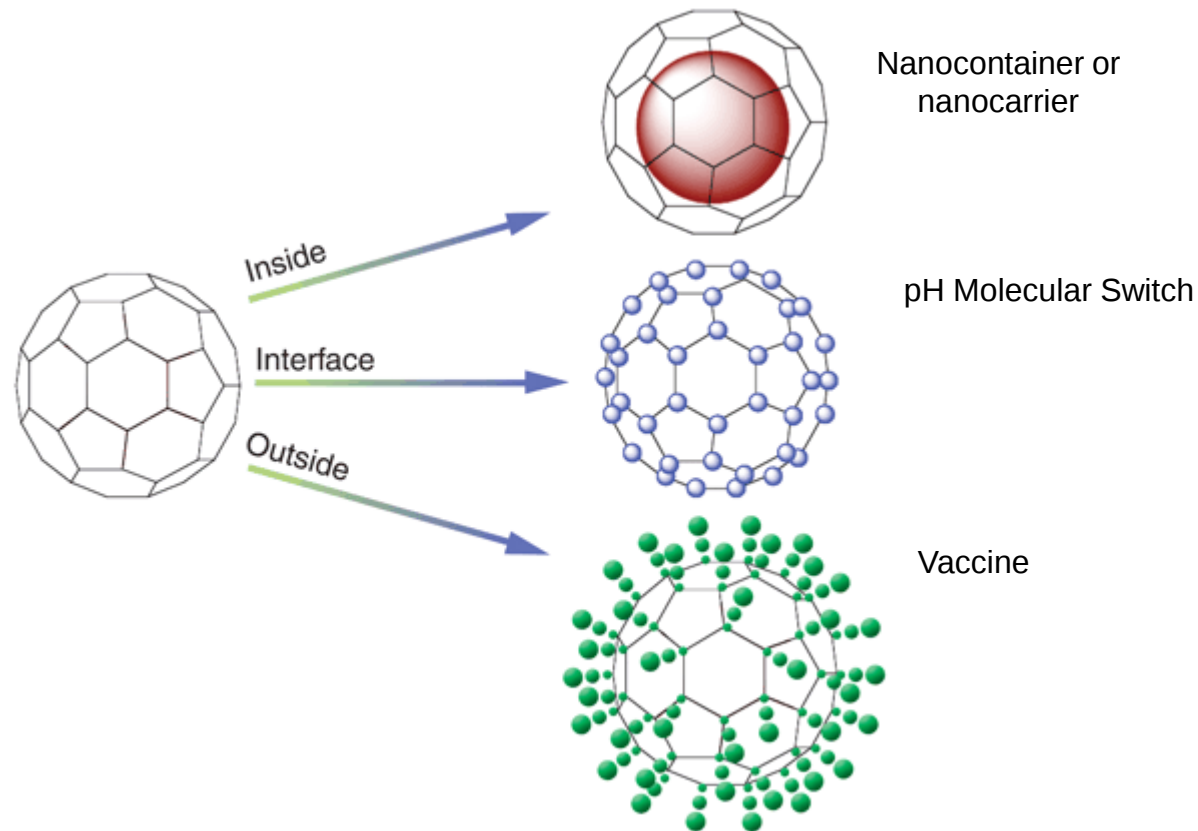
armchair edge



Biomolecules: Virus-like particles (VLP)



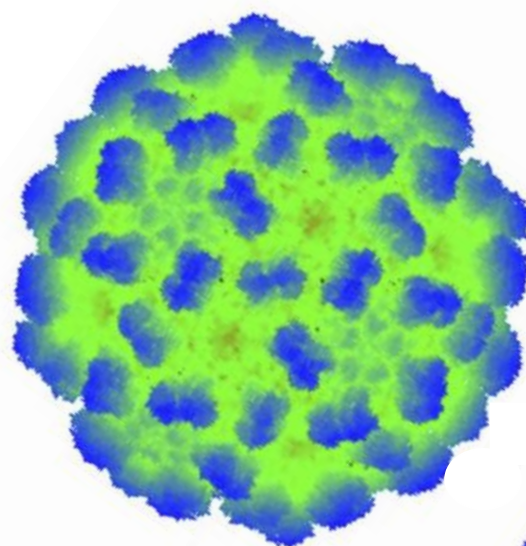
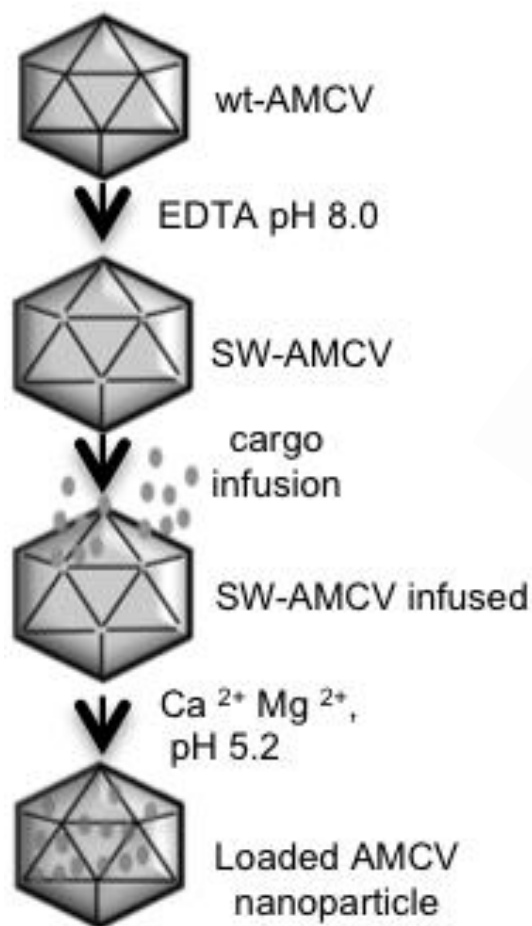
VLP can be used as platforms for synthetic and genetic manipulation with a range of applications from materials science to medicine



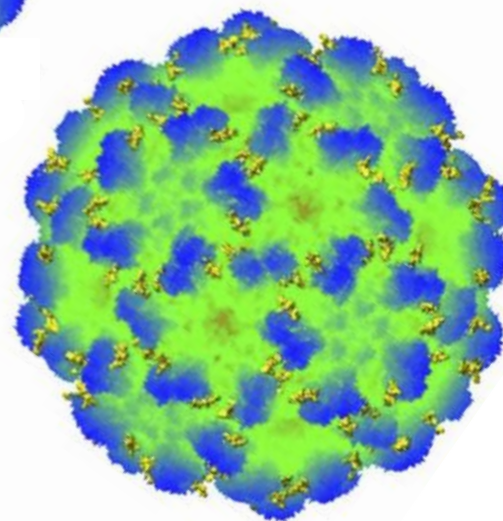
Biomolecules: Virus-like particles

Nanocarrier for drug delivery.
Swelling and filling of
doxorubicin for chemotherapy

Surface functionalization by immunogenic epitopes for immunological
response: optimization and stability criteria by MD simulations

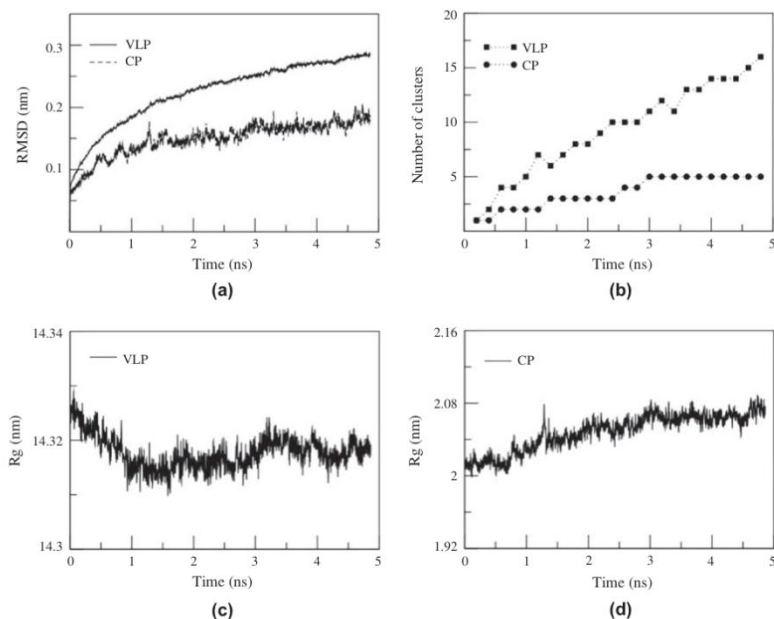


Atoms are colored on the basis of their accessibility to the solvent (red: buried; green: partially solvent exposed; blue fully solvent exposed).



The 2F5 epitope, inserted at the C-terminal of each protein of the chimeric VLP, is depicted in yellow.

Biomolecules: Virus-like particles



4 millions of atoms

GROMACS 4.5.4 on 128 cores - 0.3 ns/day

Long simulations (μ s , ms) to study the

- swelling mechanism of the VLP-based drug carriers
- the stability of the modified VLP-based vaccines
- the folding of peptides

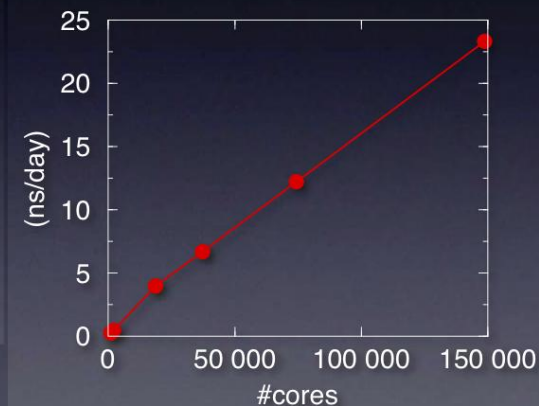
Figure 3. (a) Temporal evolution of the C- α RMSD values of the VLP (solid line) and of the single CP unit (dashed line). (b) Temporal evolution of the number of VLP clusters (solid squares) and of CP clusters (solid circles). (c) Temporal evolution of the Rg values of the VLP. (d) Temporal evolution of the Rg values of the single CP unit.

Scaling GROMACS to > 1000 cores

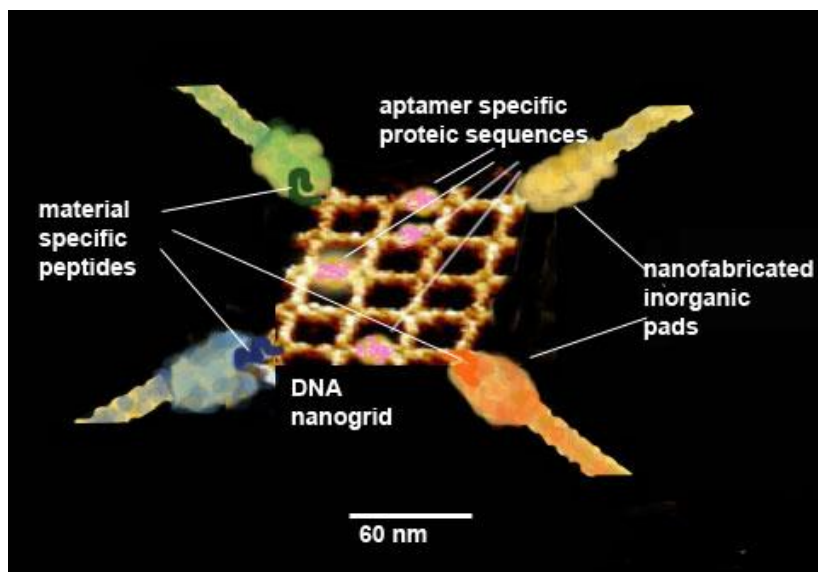
JaguarPF at Oak Ridge

- Cray XT5
- 150 000+ AMD Opteron 2.3 GHz cores
- SeaStar 2+ interconnect
- Upgrade planned to 450 000 cores

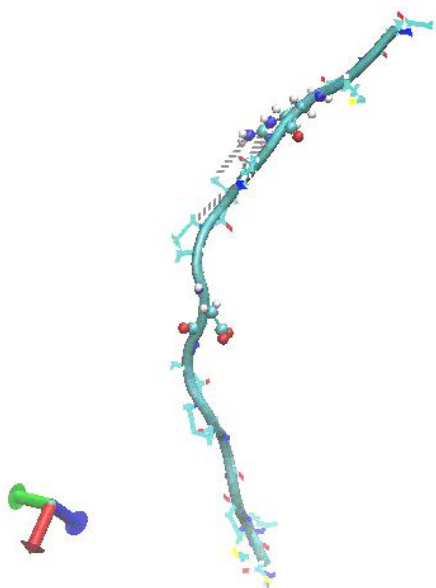
- peptides + H₂O
- 102M atoms
- Reaction-field
- 1.2 nm cut-off
- no DLB



Biomolecules: Peptides on TiO_2 surfaces



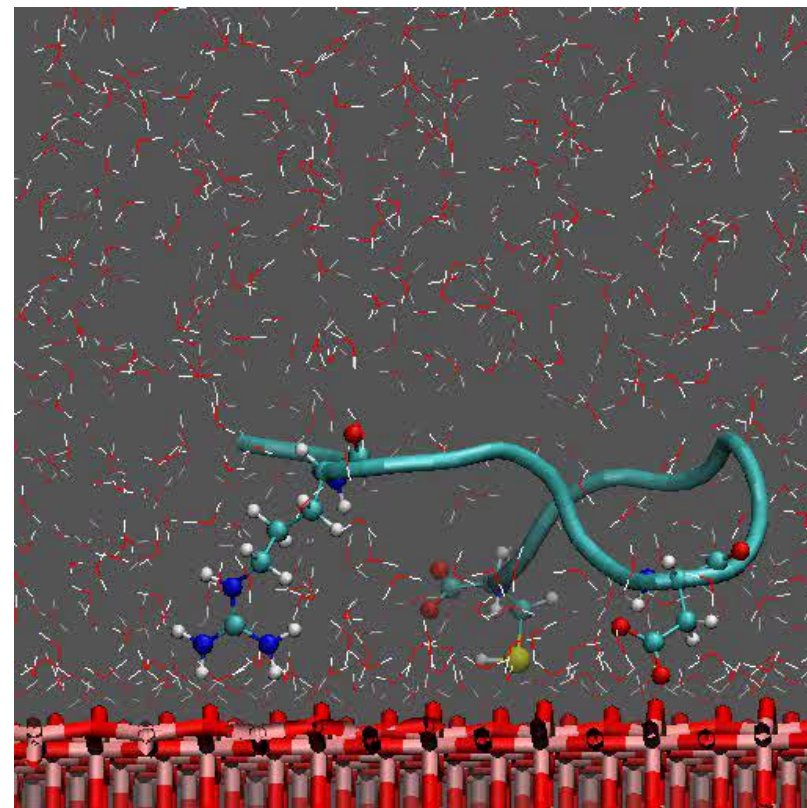
Classical Molecular Dynamics simulations are carried out to investigate both the structure and the stability of a full peptide on the (101) surface of TiO_2 in the anatase crystalline structure. Our results shed light on the role played by some amino acids that are known to be essential in selective adsorption on TiO_2 , as well as on the peptide structural conformation upon the surface. The simulations were carried on CRESCO HPC cluster by using the highly optimized parallel version of **GROMACS** (version 4.5.4).



1 microsecond trajectory
Peptide 206 atoms
8000 water molecules
100 mM NaCl
= 24771 atoms

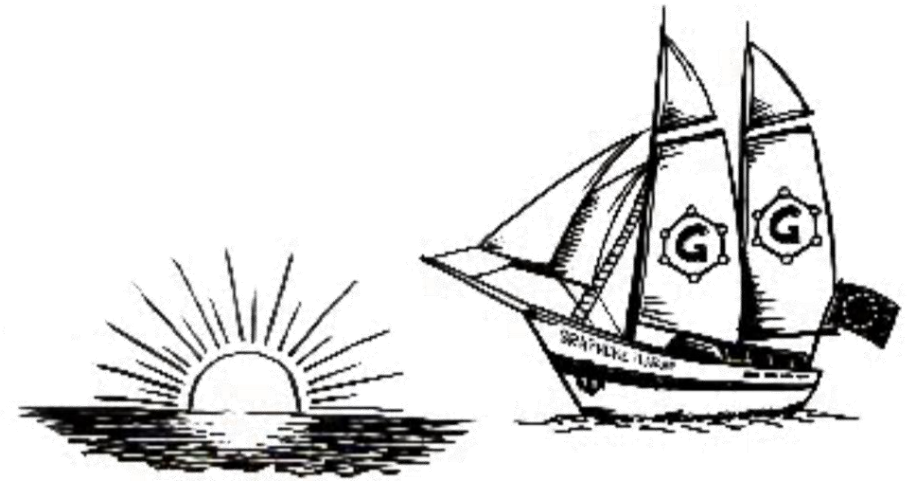
72 cores (Oak Ridge
National Labs)
performance: 63 ns/day

**FP7 Marie Curie
META Project**



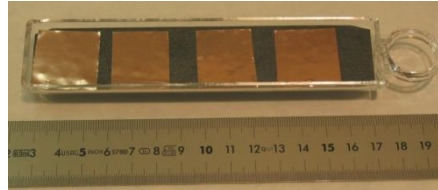
Flagship Competitive Calls

- The Scientific and Technological Roadmap forms the basis for the research program of the flagship: 11 scientific and technological work packages
- The CP-CSA (the initial flagship consortium) will include about 100 groups representing about 76 legal partners from 17 countries; additional groups will join later through competitive calls
- An open call will be published on **Dec. 2013**. About 20-30 new partners will then be selected and they are expected to enter after one year
- ERA-NET multinational calls planned, maybe in **2015**
- Horizon 2020 expansion in **2016**, expect that 150-200 partners will be included in the H2020 program
- **ENEA** with UTTMAT unit will submit to join open calls
- Before to submit, it is crucial to demonstrate our skills and know-how with publications presenting experimental reports and theoretical modeling



Nanotechnologies: atomistic model of graphene

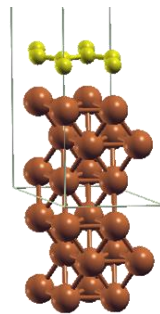
- **What** → Interfaces graphene/copper and CVD growth mechanisms



copper catalyst foils to be inserted in
chemical vapor deposition furnace

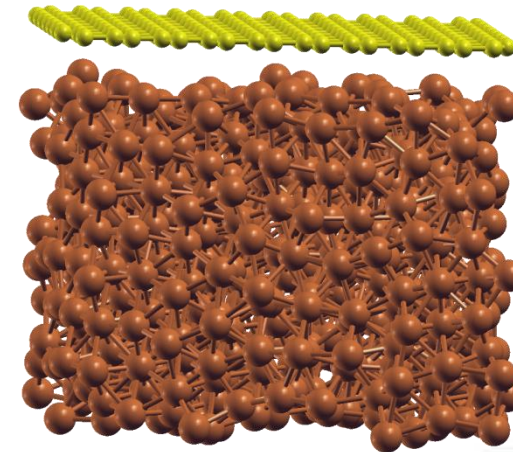
- **Why** → Quality improvement of graphene grown in CVD for photovoltaic applications; CVD low T growth
- **How** → Density functional theory (Quantum Espresso)

From

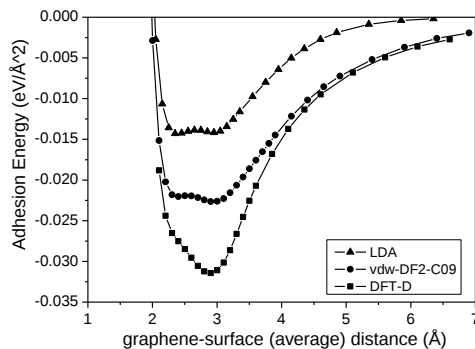


6 layers of crystalline Cu
(24 atoms) + graphene
(8 atoms)

to

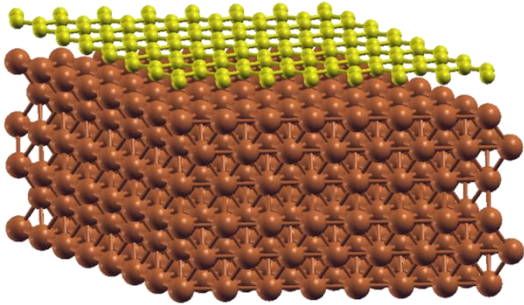


Amorphous Cu (547
atoms) + graphene
(200 atoms)



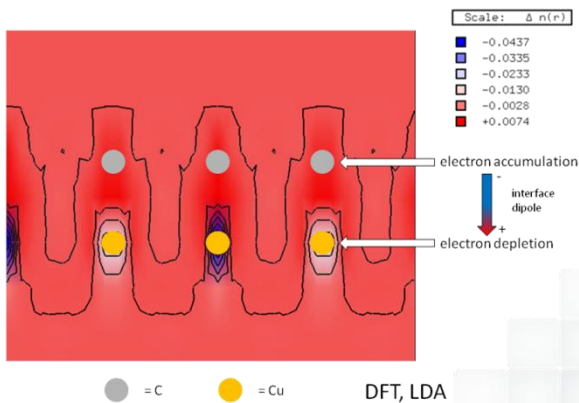
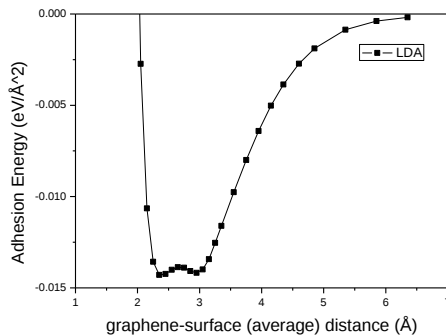
- Adhesion energies are to be calculated
- Growth mechanism models to be developed

Nanotechnologies: atomistic model of graphene

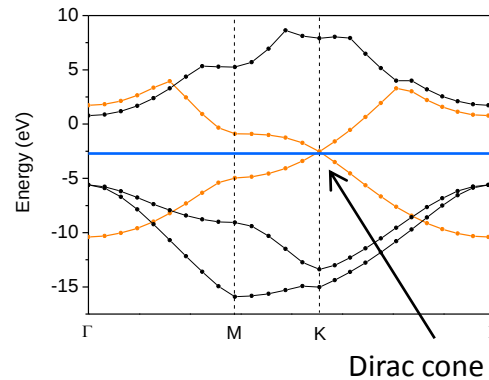


DFT simulations are based on plane waves and exchange correlation functional in local density approximation (LDA), gradient corrected approximation (GGA) also with dispersion force corrections (DFT-D, vdW-DF, vdW-DF2, C09)

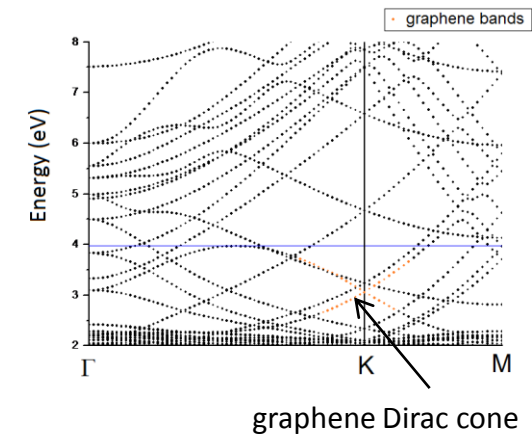
Graphene-Copper Adhesion Energy



Graphene Band Structure

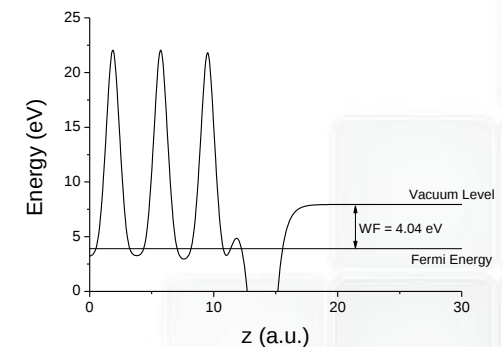


Graphene-Copper Band Structure



Charge density and interface dipole

Work Function Calculation of Copper Functionalized with Graphene



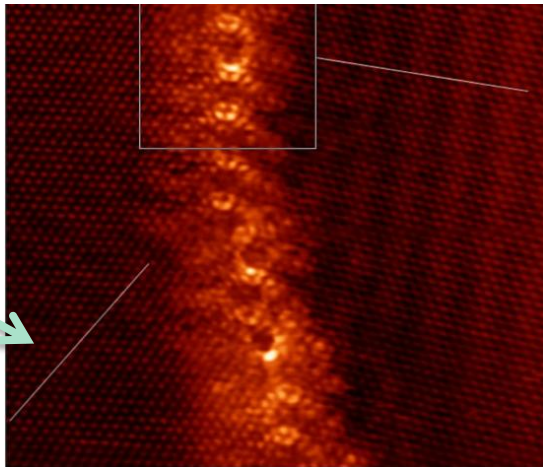
Nanotechnologies: modeling STM micrographs

Collaboration with Queensland University of Technology, Brisbane, Australia

STM micrographs of grain boundaries induced G waves

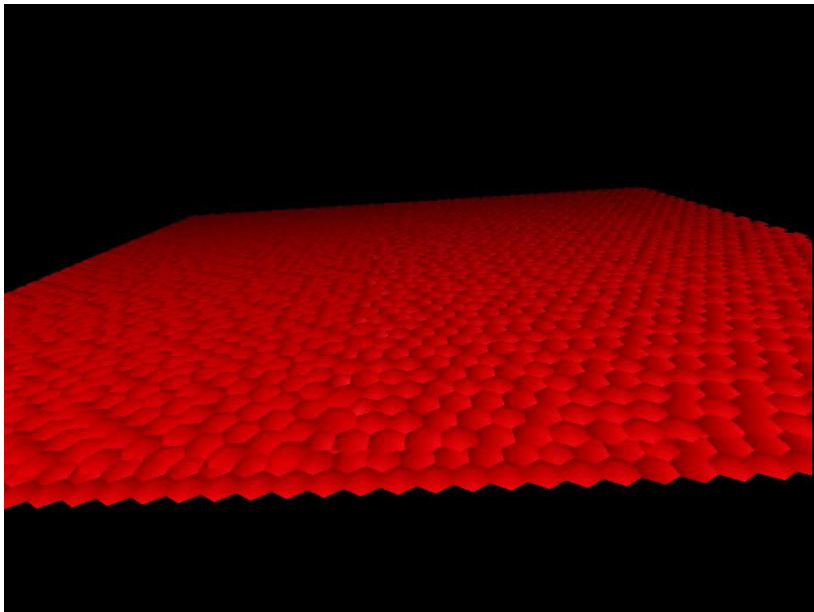
Constant-height
STM micrograph

oscillations

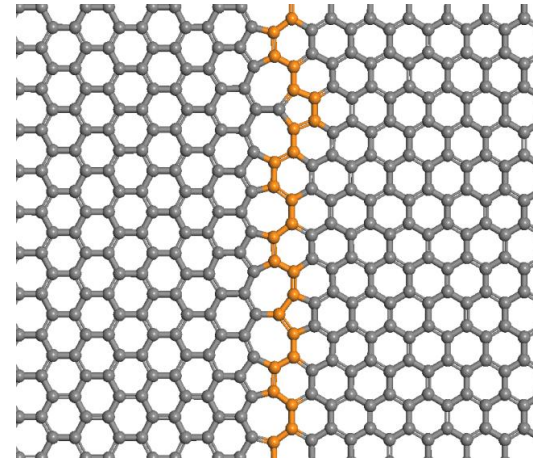


oscillations

2 graphene sheets, 10000 atoms

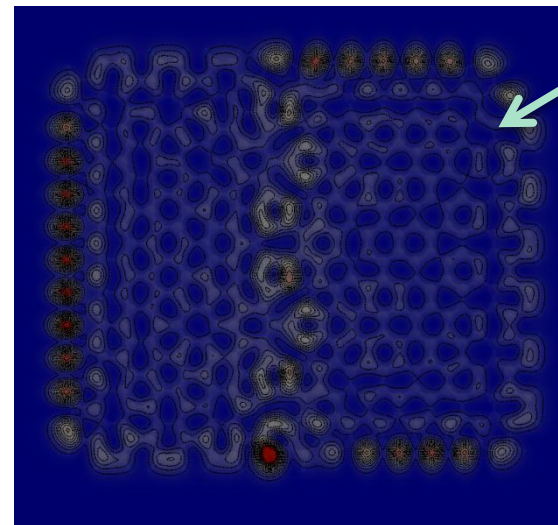


Grain boundary numerical model



zig-zag edge

armchair edge

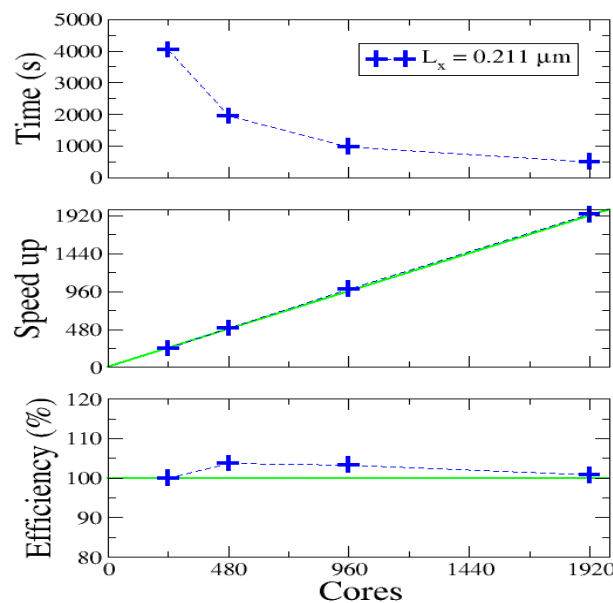
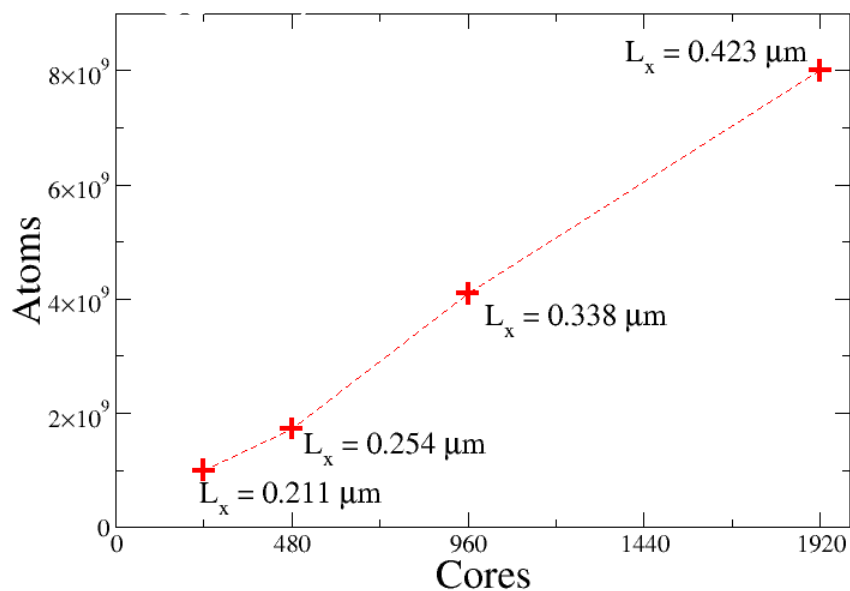
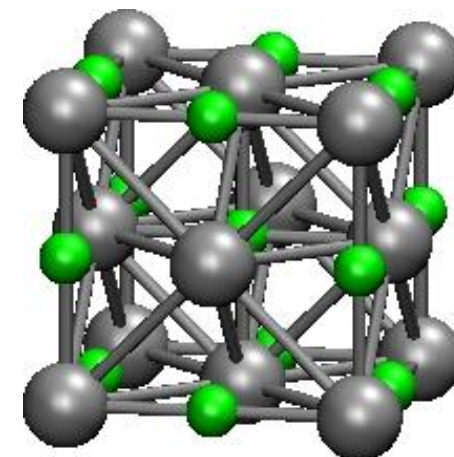


H-passivated
cluster (about
500 atoms)

STM image DFT calculated

Materials: PdH and LAMMPS

Cores	Cell	Atom _s (x10 ⁶)	L _x (μm)	Mem.* (MB)	Time** (s)	Restart (GB)
240	500x500x500	1000	0.211	2441	4043	112
480	600x600x600	1728	0.254	2103	3575	194
960	800x800x800	4096	0.338	2485	4325	458
1920	1000x1000x1000	8000	0.423	2441	4026	895
* Max. mem. for core 2667 MB; **Time for 100 MD steps						



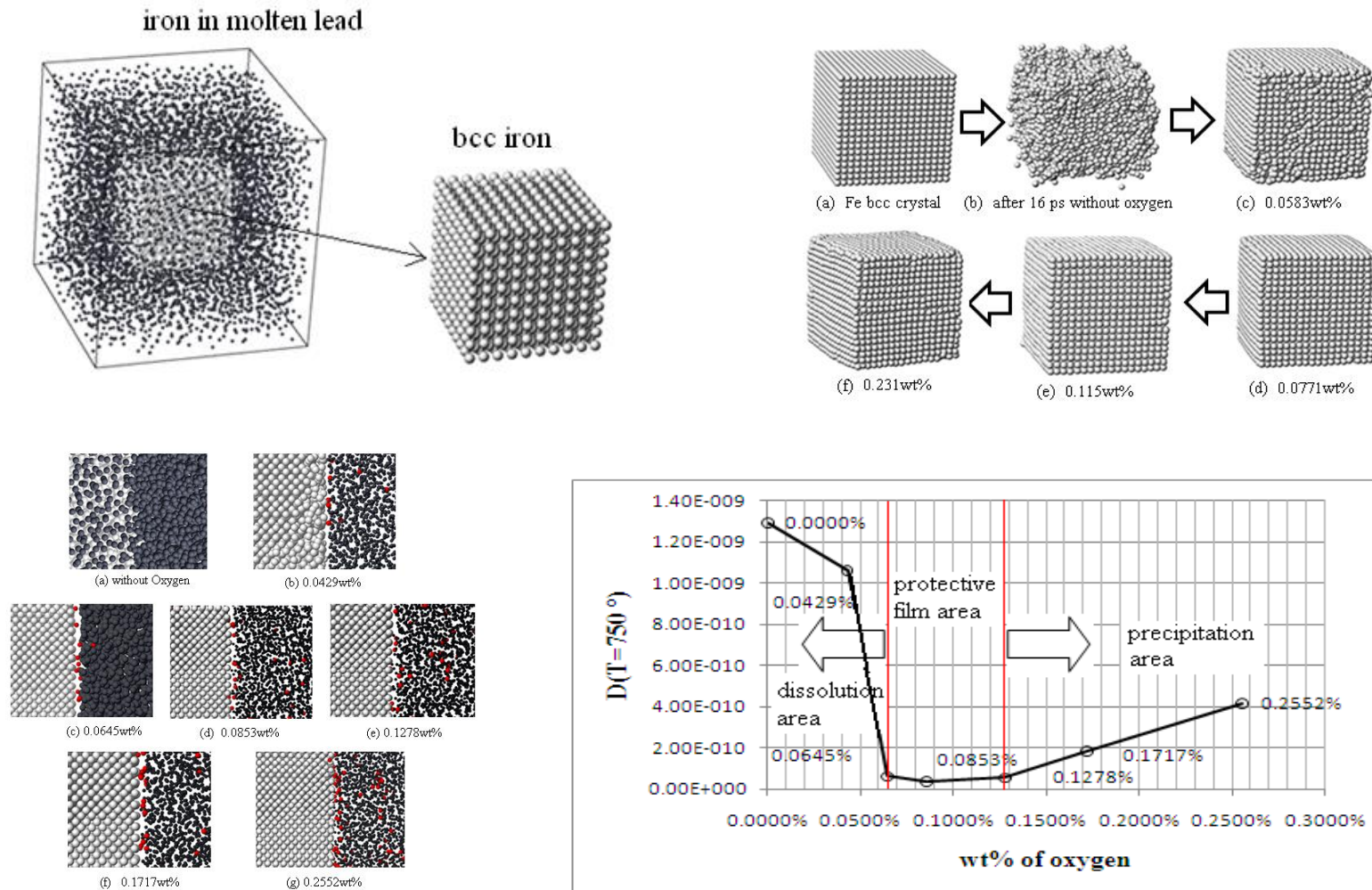
Scalability
for the
500x500x500
cells system

Materials: Iron corrosion in liquid lead

Collaboration with Indonesia and EERA – Nuclear Materials European initiative

Corrosion inhibition of iron in high temperature molten liquid lead by using oxygen injection

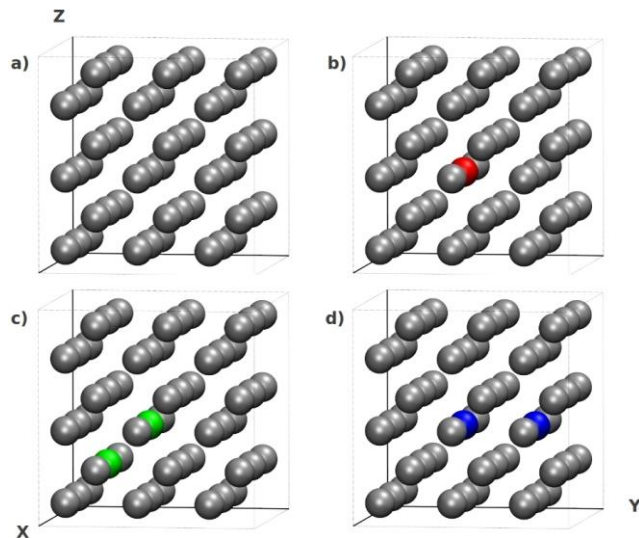
Codice MOLDY
100.000 atomi



- A.Arkundato, Z.Suud, M.Abdullah, W.Sutrisno, M.Celino, "Numerical study: iron corrosion-resistance in lead-bismuth eutectic coolant by molecular dynamics method", Int. Conf. on Advances in Nuclear Science and Engineering ICANSE2011, AIP Conference Proceedings **1448** (2012) 155-163
- A.Arkundato, Z.Suud, M.Abdullah, W.Sutrisno, M.Celino, "Molecular dynamics simulation of iron corrosion-reduction in high temperature molten lead using oxygen injection", accepted on Annals of Nuclear Materials

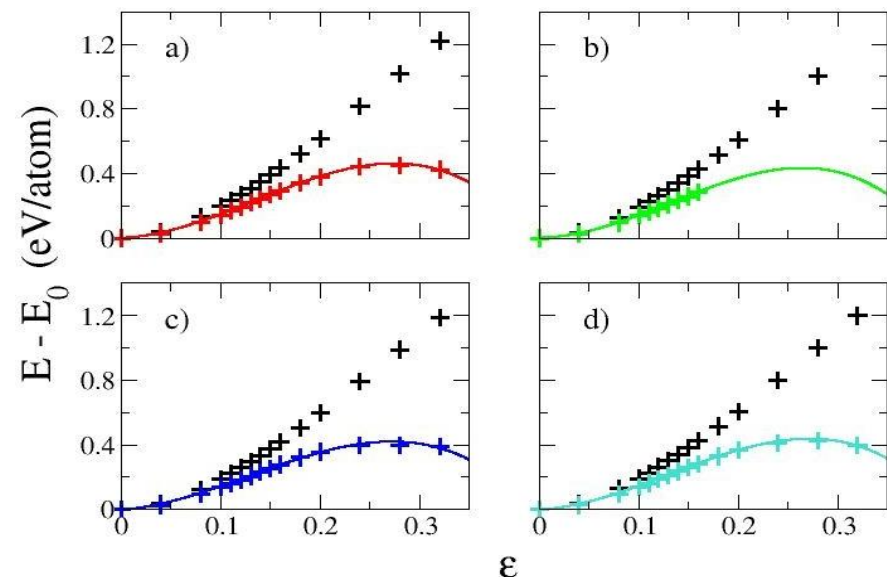
Structural and mechanical properties of tungsten and tungsten alloys (Re, Ta, V) and role of vacancies in tungsten.

- PWSCF code of **Quantum Espresso** suite;
- Supercell of **54 atoms** of W 4x4x4 bcc cell;
- *Ab initio* modelling: DFT with plane wave expansion;
- Normconserving pseudopotential with PBE exchange-correlation functional for **W, Re, H, He**.



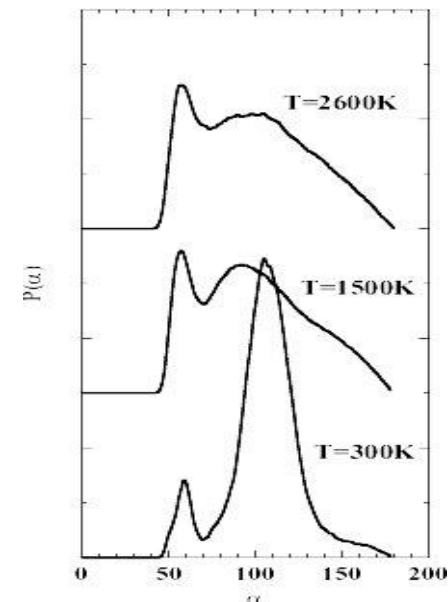
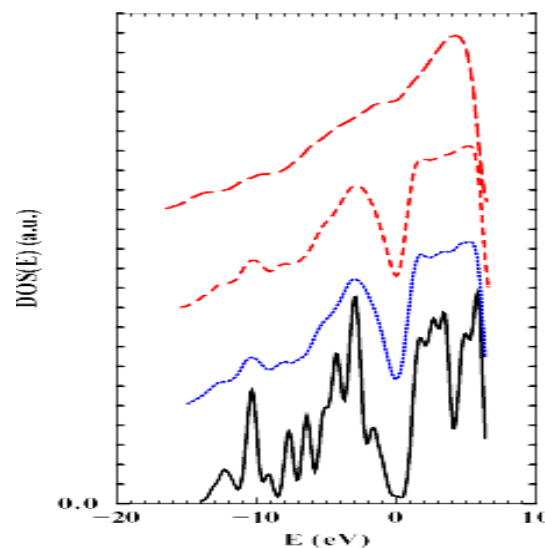
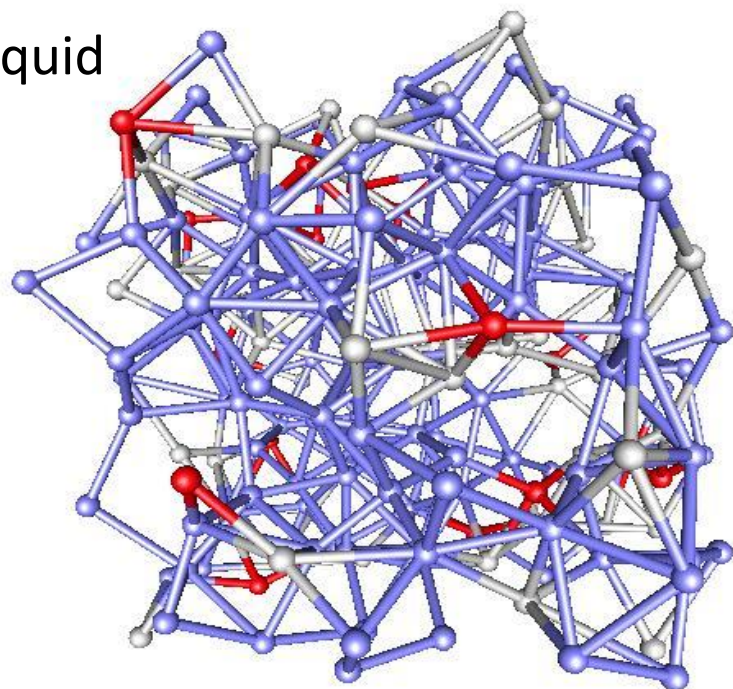
Tungsten atoms are the gray spheres. b) monovacancy case; one tungsten atom is removed (red sphere). c) divacancy 1NN case; two tungsten atoms are removed in [111] direction (green spheres). d) divacancy 2NN case; two tungsten atoms are removed in [100] direction (blue spheres).

Total energy per atom as a function of strain for uniaxial deformation. a) bcc W monovacancy case; b) bcc W divacancy 1NN case; c) and d) bcc W divacancy 2NN case. For a), b) and c) uniaxial deformation in [001] direction. For d) uniaxial deformation in [100] direction. is the ground-state energy of the systems. Energies for unrelaxed conditions are shown in black symbols, those for relaxed conditions are shown in red, green, blue, and light blue symbols respectively. Solid lines are cubic fitting functions.



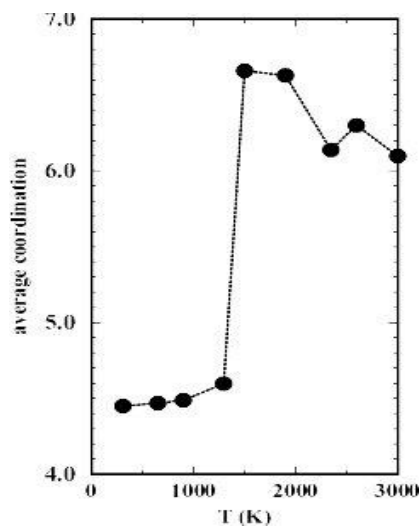
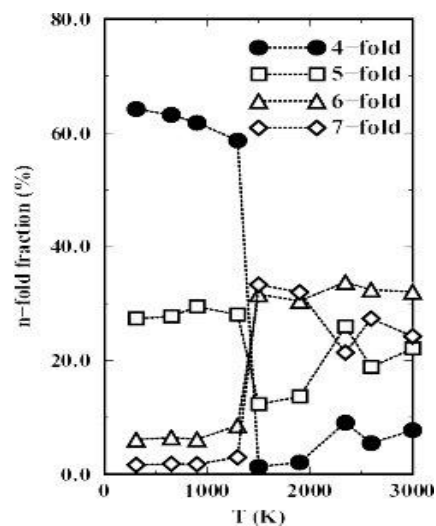
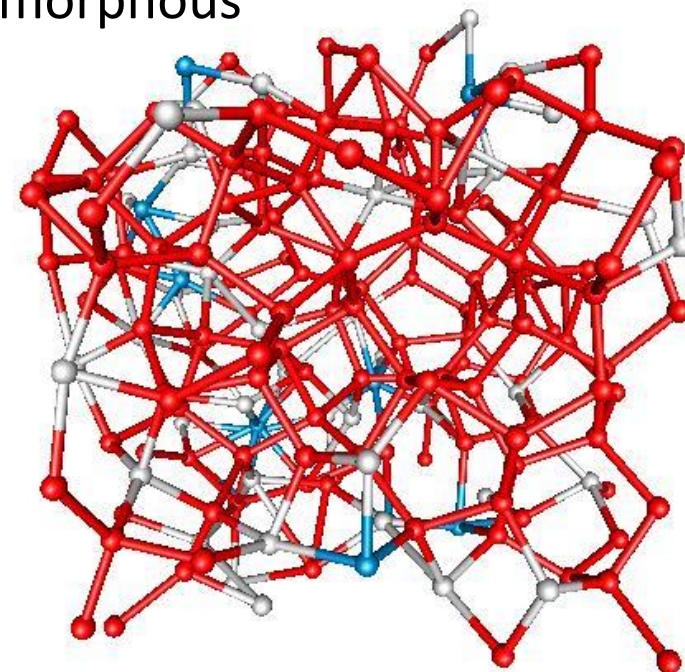
- Every point in the graph 3 day of simulations on 96 cores of crescof
- More than 120 simulations

Liquid



- 4-fold coordination
- 5-fold coordination
- 6-7-fold coordination

Amorphous



Studio delle proprietà termiche di fluidi termovettori con metodi ab-initio



Un fluido scorre in un tubo lungo la linea focale del concentratore; le proprietà termiche e la stabilità chimica del fluido sono di fondamentale importanza

Impianto PCS (ENEA): miscela di nitrati di Sodio e Potassio

Obiettivi:

- studiare il fluido termovettore con metodi ab-initio (capacità termica, conduttività, viscosità, densità in funzione della T)
- ottenere indicazioni su possibili variazioni della miscela o aggiunta di componenti eterogenee, come nanoparticelle
- stabilità chimica e interazione con altri materiali (corrosione o degrado del fluido)

Materials: thermal properties of fluids

Punti di fusione:

NaNO_3 : 308 °C

KNO_3 : 334 °C

Miscela in uso su PCS : 230 °C

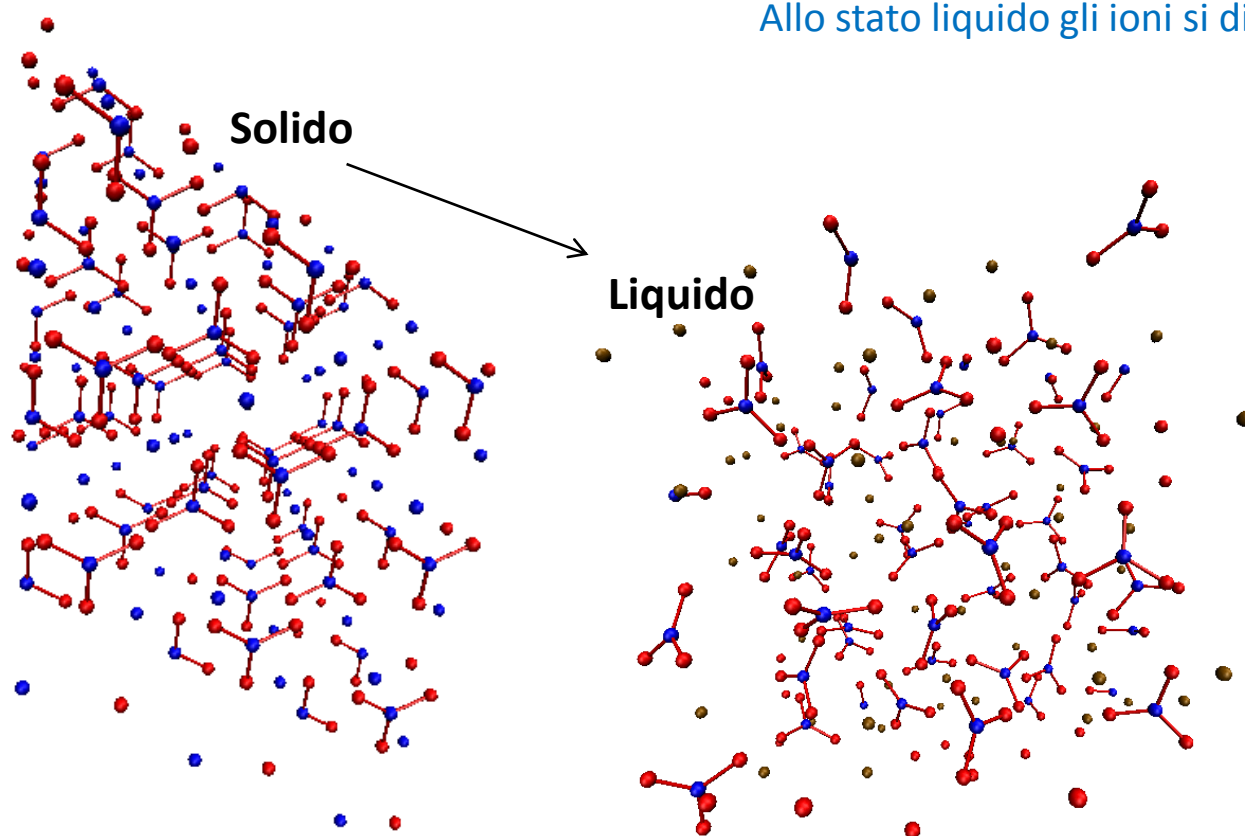
Sali parzialmente miscibili (formano un solido eterogeneo)

La miscela eutettica ha rapporto molare 50% / 50% (NaNO_3 / KNO_3)

Il sale in uso su PCS ha rapporto molare 64% / 36% (NaNO_3 / KNO_3)

Solidi ionici: gruppi NO_3^- e Na^+/K^+

Allo stato liquido gli ioni si dissociano e diventano conduttori elettrici



Simulazione di una cella con 270 atomi (corrispondenti a 9 celle cristalline elementari)

Risorse usate per un calcolo tipico: 144 cores

Tempi di calcolo: 24 ore $\rightarrow$ 0.1 - 0.2 ps

Proprietà microscopiche del liquido $\rightarrow$ 2-3 ps

Proprietà macroscopiche $\rightarrow$ 5-6 ps

Per ottenere risultati sui cambiamenti di fase sono richieste simulazioni di qualche decina di ps su un sistema più grande (solido in equilibrio con il liquido)